817

## A Convenient Synthesis of 7-Methoxymitosene by the Photolysis of Aminobenzoquinones

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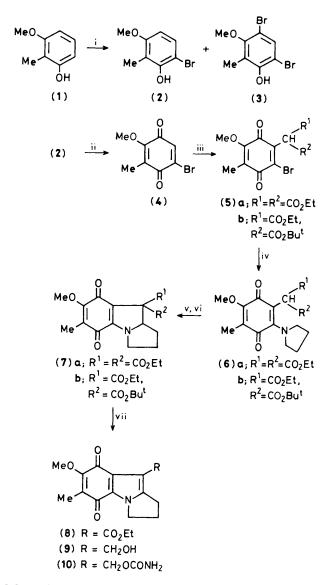
A simple method for the preparation of pyrrolo[1,2-*a*]indoloquinone derivatives using the photolysis of amino-quinones has led to a formal synthesis of 7-methoxymitosene.

In synthetic studies on mitomycin antibiotics,<sup>1</sup> which contain several unique structural features, we have shown that the photolysis of amino-1,4-naphtho- and -benzo-quinones having the active methylene group at the 2-position provides a preparative route to heterocyclic quinones.<sup>2</sup> In this communication we describe the application of this photo-induced reaction to freshly prepared pyrrolidinylbenzoquinones (6) as a simple synthesis of 7-methoxymitosene (10).

Our initial approach to (10) focused on the preparation of

(7) as outlined in Scheme 1. Treatment of the phenol (1)<sup>3</sup> with bromine-t-butylamine<sup>4</sup> at -78 °C afforded the bromophenol (2) (60%, m.p. 47 °C) in addition to the dibromophenol (3) (20%, m.p. 74 °C). Compound (2) was oxidized with potassium nitrosyldisulphonate-KH<sub>2</sub>PO<sub>4</sub>( $\frac{1}{6}$  M)-acetone-H<sub>2</sub>O to give the quinone (4)† (73%, m.p. 65 °C). Treatment of (4) with

† Satisfactory spectral and analytical data were obtained.



Scheme 1. Reagents: i, Br<sub>2</sub>, Bu<sup>t</sup>NH<sub>2</sub>, -78 °C; ii,  $(SO_3K)_2NO$ , 1/6 M KH<sub>2</sub>PO<sub>4</sub>, Me<sub>2</sub>CO-H<sub>2</sub>O; iii, TlCH(R<sup>1</sup>)(R<sup>2</sup>) (R<sup>1</sup> = R<sup>2</sup> = CO<sub>2</sub>Et or R<sup>1</sup> = CO<sub>2</sub>Et, R<sup>2</sup> = CO<sub>2</sub>Bu<sup>t</sup>), THF; iv, pyrrolidine, CHCl<sub>3</sub>; v, hv, EtOH; vi, SiO<sub>2</sub>, EtOH; vii, CF<sub>3</sub>CO<sub>2</sub>H, CH<sub>2</sub>Cl<sub>2</sub>.

thallium diethylmalonate in tetrahydrofuran (THF) gave  $(5a)^{\dagger}$  (40%, a yellow oil), which reacted with pyrrolidine to yield the amino-quinone (6a)<sup> $\dagger$ </sup> (85%, a purple oil). A solu-

tion of (6a) in ethanol was irradiated with a high pressure mercury lamp through Pyrex glass. This irradiated solution was retained on a silica gel column for a few days, and then eluted with ethyl acetate to afford (7a)<sup>+</sup> (68%, m.p. 171 °C). De-esterification of (7a) with sodium cyanide or magnesium chloride in dimethyl sulphoxide5 followed by oxidation failed to give the required product (8). Therefore, the pyrrolidinylbenzoquinone (6b)<sup>†</sup> was prepared from (4) and the thallium salt of ethyl t-butylmalonate by the same method in a moderate yield. Photolysis of (6b) afforded the diastereoisomers of the pyrroloindoloquinone, (7b)-(I)<sup>+</sup> (40%, m.p. 103 °C) and (7b)-(II)  $\ddagger$  (32%, m.p. 156 °C), in the ratio 5:4 after chromatography on silica gel. Structural assignments have not yet been made. On treatment with trifluoroacetic acid, however, each stereoisomer was converted into the same product (8)‡ (m.p. 164 °C), which was identical with an authentic sample (lit.<sup>6</sup> m.p. 165-166 °C). Recently, the transformation of (8) to (9) and 7-methoxymitosene (10) was reported by Coates and MacManus.<sup>6</sup> Consequently, this sequence constitutes a formal synthesis of 7-methoxymitosene.

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## References

- 1 R. W. Franck, *Prog. Chem. Org. Nat. Prod.*, 1979, **38**, 1; K. Takahashi and T. Kametani, *Heterocycles*, 1979, **13**, 411; J. R. Lury and H. Rapoport, *J. Org. Chem.*, 1982, **47**, 2404. Recent references are contained herein.
- 2 M. Akiba, Y. Kosugi, M. Okuyama, and T. Takada, J. Org. Chem., 1978, 43, 181; M. Akiba, S. Ikuta, and T. Takada, Heterocycles, 1981, 16, 1579.
- 3 A. Rashid, J. Chem. Soc., 1967, 1323.
- 4 D. E. Pearson, J. Org. Chem., 1967, 32, 2358.
- 5 W.S.Johnson, C.A. Harbert, and R. D. Stpanovic, J. Am. Chem. Soc., 1968, 90, 5279.
- 6 R. M. Coates and P. A. MacManus, J. Org. Chem., 1982, 47, 4822.

‡ Compound (7a): i.r.  $v_{max}$ (KBr) 1720, 1650 (C=O) cm<sup>-1</sup>; m/z377  $M^+$ ; <sup>1</sup>H n.m.r.  $\delta$ (CDCl<sub>3</sub>) 1.30 (6H, t, J 7.5 Hz, Me × 2), 1.87 (3H, s, Me), 1.70–2.20 (4H, m, CH<sub>2</sub> × 2), 3.63 (2H, m, CH<sub>2</sub>N), 4.12 (3H, s, OMe), 4.26 (4H, q, J 7.5 Hz, OCH<sub>2</sub> × 2), 4.80 (1H, m, CHN). Compound (7b)–(1): i.r.  $v_{max}$  (KBr) 1742, 1723, 1660, 1640, 1580 cm<sup>-1</sup>; m/z 405  $M^+$ ; <sup>1</sup>H n.m.r.  $\delta$ (CDCl<sub>3</sub>) 1.28 (3H, t, J 7.5 Hz, Me), 1.46 (9H, s, Bu<sup>t</sup>), 1.84 (3H, s, Me), 1.84–2.20 (4H, m, CH<sub>2</sub> × 2), 3.60 (2H, m, NCH<sub>2</sub>), 4.08 (3H, s, OMe), 4.26 (2H, q, J 7.5 Hz, CH<sub>2</sub>), 4.80 (1H, m, CH). Compound (7b)–(II): i.r.  $v_{max}$  (KBr) 1747, 1708, 1650, 1623, 1560 cm<sup>-1</sup>; m/z 405  $M^+$ . The n.m.r. spectrum is very similar to that of (7b)–(1). Compound (8): i.r.  $v_{max}$  (KBr) 1718, 1662, 1640, 1610 cm<sup>-1</sup>; m/z 303  $M^+$ ; <sup>1</sup>H n.m.r.  $\delta$ (CDCl<sub>3</sub>) 1.36 (3H, t, J 7.5 Hz, Me), 1.96 (3H, s, Me), 2.56 (2H, quintet, J 7.5 Hz, NCH<sub>2</sub>CH<sub>2</sub>CH<sub>2</sub>), 3.10 (2H, t, J 7.5 Hz, CH<sub>2</sub>), 4·06 (3H, s, OMe), 4.28 (2H, t, J 7.5 Hz, NCH<sub>2</sub>), 4.32 (2H, q, J 7.5 Hz, CO<sub>2</sub>CH<sub>2</sub>Me).